

Findings:

pH 5: +ve charges on ferrihydrite

- Fh⁺ + G3.5-COO⁻ → Fh-G3.5-COO
 - electrostatic bonding is dominant
- G4-OH & unt-NGQD do not deprotonate
 - non-charge bonding

pH 7 → pH 11: gradually less +ve charges

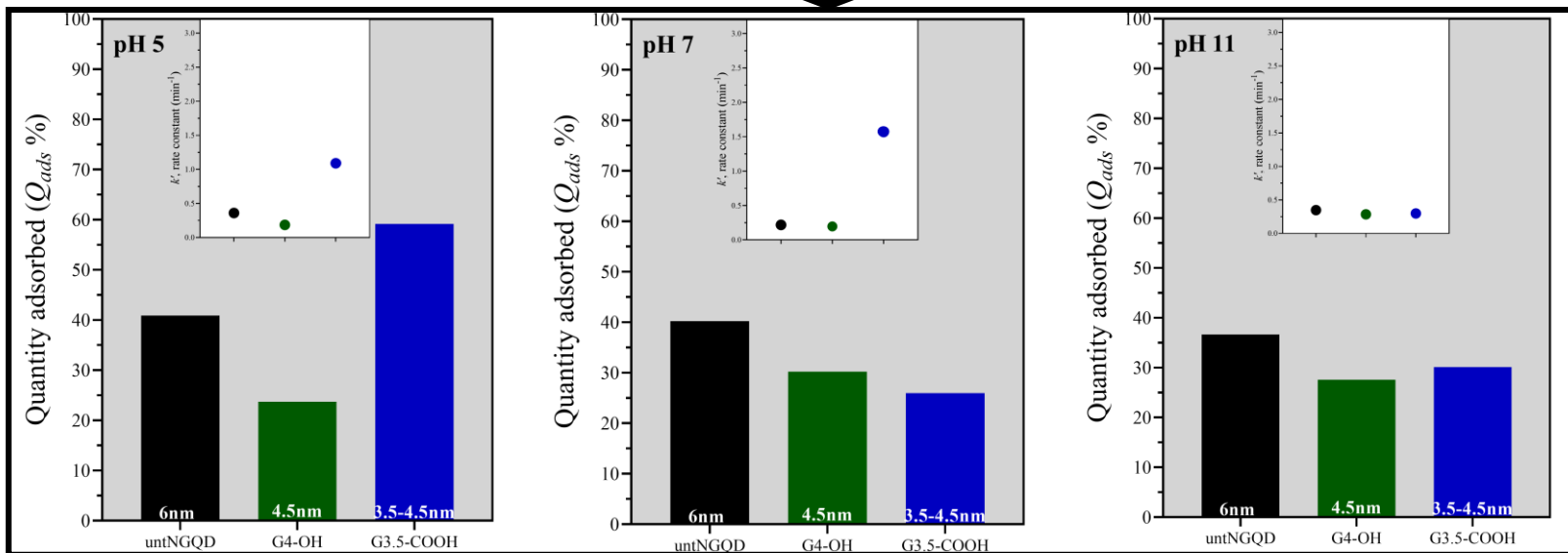
- less +ve sites for G3.5-COO⁻ to bind to
- slower reactions = non-charge bonding

G3.5-COOH behavior: pH dependent

G4-OH & untGQDs behavior: no evidence of pH dependence

Applications:

- Organics behavior in earth's systems
 - Mechanism and kinetics
- Model for other studies involving organics and minerals
- Long term applications: environmental remediation, curbing climate change



ORGANIC NANOMATERIALS-FERRIHYDRITE INTERACTIONS

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