







Energetics and Binding Dynamics of Natural Organic Matter (NOM) with Iron (III) Hydr(Oxides) as Studied by Flow Adsorption Microcalorimetry

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		Microbially-processed (NOM)			Fire-processed (AJ450)	
		pН	Sorption	Desorption	Sorption	Desorption
	Rate (min ⁻¹)					
		3	1.81 ± 0.2	0.93 ± 0.001	0.28 ± 0.001	0.82 ± 0.008
		5	3.88 ± 0.01	1.00 ± 0.009	0.37 ± 0.008	1.16 ± 0.008
Ĩ		7.5	3.18 ± 0.7	1.60 ± 0.002	0.93 ± 0.1	1.67 ± 0.14
	Energy $(J g^{-1})$					
		3	0.19 ± 0.1	0.13 ± 0.02	0.24 ± 0.2	0.11
		5	0.09 ± 0.02	0.07 ± 0.01	0.22 ± 0.01	0.09
		7.5	0.06 ± 0.1	0.06 ± 0.02	0.06 ± 0.02	0.071
ור	Q _{ad/des} (mg/g)					
;		3	61.82	22.25	49.23	35.14
5		5	83.14	56.53	58.07	12.62
6 0		7.5	28.32	19.82	49.91	25.83
00	$\Delta H (kJ/g*10^{-3})$					
		3	3.07	5.84	4.87	3.13
		5	1.08	1.27	3.78	7.13
		7.5	2.12	3.17	1.2	2.71

- Rate:
- NOM faster than AJ450
- Desorption is comparable
- Increase with increasing pH
- Energy:
 - AJ450 more energetic than NOM
 - Desorption is comparable
 - Decrease with increasing pH
- ΔH:
- Varies with pH
- Generally higher at pH 3

		Microbially-pro	cessed (NOM)	Fire-processed (AJ450)	
	pН	Pre	Post	Post	
Rate (min ⁻¹)					
	3	0.82 ± 0.005	0.68 ± 0.009	0.87 ± 0.03	
	5	0.86 ± 0.01	0.94 ± 0.004	1.05 ± 0.02	
	7.5	1.01 ± 0.005	0.96 ± 0.02	1.02 ± 0.01	
Energy $(J g^{-1})$					
	3	7.80 ± 0.8	8.75 ± 0.15	7.50 ± 0.6	
	5	2.90 ± 0.1	2.40 ± 0.1	1.04 ± 0.2	
	7.5	0.75 ± 0.05	1.90 ± 0.1	0.65 ± 0.05	

- Rate:
- Slight decrease for NOM and increase for AJ450
- Increase with decreasing pH
- Energy:
 - Slight increase for both AJ450 and NOM
 - Decrease with increasing pH