Two-Dimensional Metal Halide Perovskites containing Triazine based Macrocycles William L. Burnett, Alexander J. Menke, Eric E. Simanek PhD. and Jeffery L. Coffer PhD.

II. Experimental

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I. Introduction

Metal halide perovskites (MHPs) are an emerging type of semiconducting crystal with the structure ABX₃ made up of an A site positively charged cation such as methylammonium (MA), B site metal cation such as lead, and anion X which is a halide such as bromide (Fig. 1).¹ MHPs exhibit important properties such as the ability to produce light when stimulated by electricity known as a light emitting diode (LED).²

Previous research has shown incorporation of large organic molecules into the MHP crystal can increase the longevity of the LED by making them more resistant to degradation caused by moisture and air.³ This work will investigate the incorporation of triazine based macrocycles to improve the light emission by forming 2D layers on 3D bulk crystal resulting in narrowing the emission and/or increasing the emission intensity.



A. 3D MHP

B. 2D MHP

C. 2D-3D MHP

Figure 1. A) Bulk 3D MHP crystal representation with lead-bromide as the green octahedra with interior A-site as the dark green circle. B) Example of 2D perovskite with the blue oval representing the macrocycle with extended layers of octahedra N = 1 and N = 2. C) 2D-3D mixture with the 2D layer forming on top of the 3D crystal.

II. Experimental

A. Thin Film Preparation

TiO₂ precursor



0.5M TiO₂ diisopropoxide bis(acetylacetonate) in

1.5M MABr + PbBr₂ in DMF/DMSO (4/1 v/v)

MAPbBr₃ precursor



Spincoat TiO₂ precursor on Fluorine doped tin oxide substrate 3000rpm 40seconds



isopropanol



Place under Vacuum overnight before analysis

Anneal 95C for 30 minutes

Spincoat Anisole 10 seconds after MAPbBr₃ precursor

Spincoat MAPbBr₃ precursor 3000rpm 40seconds



A. Macrocycle Preparation a) Boc-Hydrazine b) Lysine or Valine 3.3-diethoxypropylamine HBTU, HOBT

TFA

Figure 2. Macrocycle Synthetic scheme. Two macrocycles were prepared and used different amino acids to form different R groups, one amine terminated (K-K) , one isopropyl terminated (V-V).

III. Results

A. Perovskite Thin Films (- Macrocycle)





Figure 3. Photoluminescent (PL) image of MAPbBr₃ thin film.



Figure 5. PL spectra of MAPbBr₃ thin film from multiple locations on substrate. Average lambda max of 537 (\pm 0.9) nm Average FWHM 33.6 (\pm 4.0) nm

Figure 4. PL image of MAPbBr_{3-x}Cl_x thin film. MACl is excess in precursor solution.

	120000
Counts Per Second	100000
	80000
	60000
	40000
	20000
	(

Acknowledgments

TCU's Department of Chemistry and Biochemistry Robert A. Welch Foundation









Figure 6. PL spectra of MAPbBr_{3-x}Cl_x thin film from multiple locations on substrate showing 2 modes of light emission. Primary lambda max of 524 (\pm 1.6) nm Average FWHM 30.1 (\pm 3.9) nm



Figure 7. PL image of MAPbBr₃ + K-K thin film.



Figure 9. PL spectra of MAPbBr₃ + K-K thin film from multiple locations on substrate. Average lambda max of 536 (\pm 0.45) nm Average FWHM 23.6 (\pm 0.8) nm



Figure 11. Powder x-ray diffractogram of thin films showing peak locations consistent with Cubic Pm3m phase and line widths consistent with high crystallinity material.

IV. Conclusions

- and more thin film like morphology.

V. Future Work

VI. References

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- dimensionality 2D/3D heterojunctions. *Science* **2022**, *376* (6588), 73–77,

Figure 8. PL image of MAPbBr_{3-x}Cl_x + V-V thin film.

 $MAPbBr_{3-x}Cl_x + V-V$



Figure 10. PL spectra of MAPbBr_{3-x}Cl_x + V-V thin film from multiple locations on substrate. Primary lambda max of 511 (± 3.5) nm Average FWHM 34.3 (± 3.4) nm

Addition of V-V to MAPbBr_{3-x}Cl_x increases the separation of the two light emission peaks enhancing halide segregation and reducing the crystal size. Addition of K-K to MAPbBr₃ showed a decrease in the FWHM of the emission

Step-wise addition of the Macrocycle to form 2D layers on 3D crystal Small angle X-ray diffraction to interrogate the possible 2D layers

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