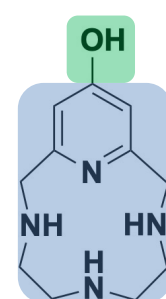


## INTRODUCTION

### FREE RADICAL SCAVENGERS



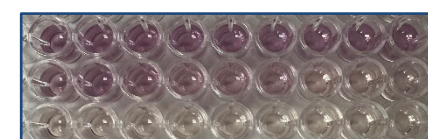
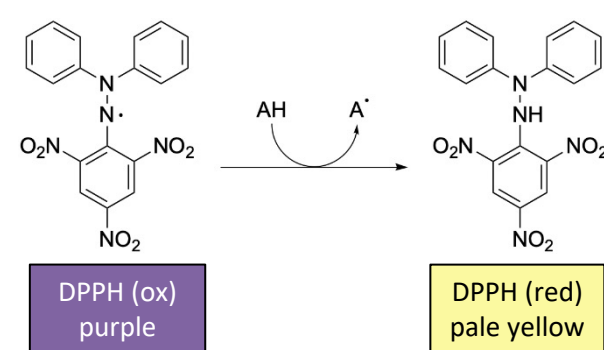
- Reactive oxygen species (ROS): Atoms/molecules that contain unpaired valence electrons and at least one oxygen within their structure.
- Necessary: Natural metabolic byproduct and serve important functions in cellular processes, including homeostasis and cell signaling.
- Mis-regulation: Severe damage to cells and membranes, can lead to neurological diseases like Alzheimer's Disease (AD).
- Radical scavengers: Substances that react with ROS to neutralize them and make them less reactive.
- Natural Antioxidant Sources Include:



Synthetic Antioxidant  
OH-PyN<sub>3</sub>

## DPPH• ASSAY

**DPPH (2,2-diphenyl-1-picrylhydrazyl) Assay:** Used to predict antioxidant activity through free radical scavenging capacity.



Main Questions to be Addressed:

1. Do the synthetic antioxidant molecules react with DPPH to demonstrate radical scavenging reactivity?

2. IF YES, what is each molecule's effect on DPPH at various concentrations?

## INITIAL PROCEDURE

Make Stock Solution of Antioxidant Molecule

Dilute Solutions to 2x the Target Concentrations (Working Solutions)

Insert Concentrated Solutions into Well Plate (Triplicates)

Quickly Add DPPH to Solutions (Rows C-E)

	1	2	3	4	5	6	7	8	9	10	11	12
A	X	X	X	X	X	X	X	X	X	X	X	X
B	B	B	B	B	B	B	B	B	B	B	B	B
C	X	NC	NC	NC	50	50	50	100	100	100		
D	X	250	250	250	500	500	500	1000	1000	1000		
E	X	1500	1500	1500	1750	1750	1750	2000	2000	2000		
F	X											
G	X											
H	X	X	X	X	X	X	X	X	X	X	X	X

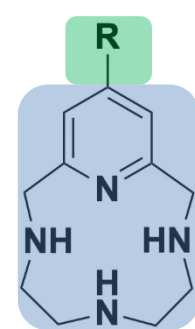
B = Blank (200  $\mu$ L Methanol)

NC = Negative Control

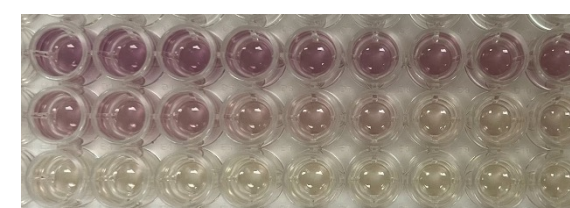
Values = Concentration of Radical Scavenging Molecule ( $\mu$ M)

Incubate for 30 min in aluminum foil & measure the absorbance (520 nm)

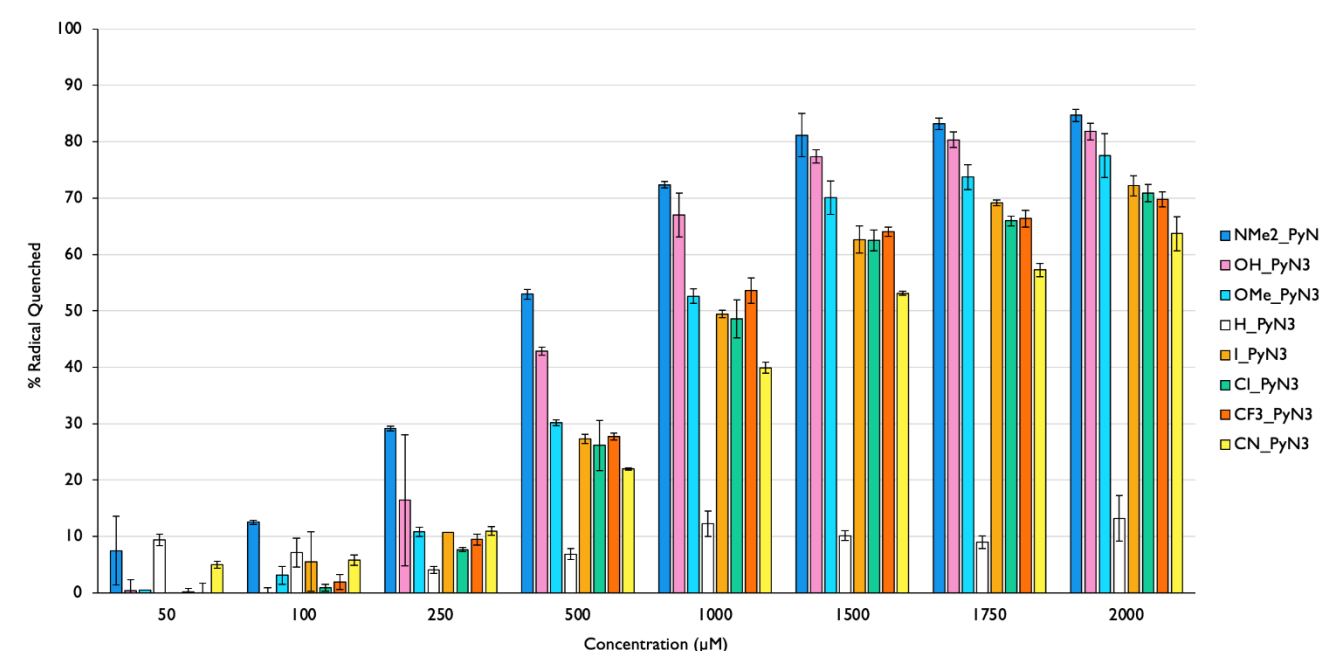
## PyN<sub>3</sub> Series



R = NMe<sub>2</sub>, OH, OMe, H, I, Cl, CN, CF<sub>3</sub>



**Design Strategy:** Pyridine rings are known to stabilize radicals as a result of  $\pi$ -electron delocalization and stabilization of radical species or the captodative effect.

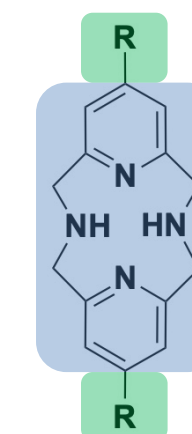


- ✓ H-PyN<sub>3</sub> is a weak radical scavenger
- ✓ At high concentrations, the substituted PyN<sub>3</sub> series shows sufficient quenching activity
- ✓ Electron-donating R groups show higher levels of scavenging radicals than electron-withdrawing substitutions
- ✓ **Conclusion:** Functionalizing the R group on the PyN<sub>3</sub> moiety increases radical scavenging activity compared to the H-PyN<sub>3</sub> parent molecule

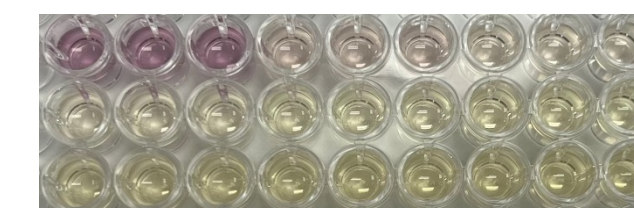
## FUTURE DIRECTIONS

- Complete the Coumarin Carboxylic Acid (CCA) assays for both series
- TCU Biology Collaborators: Cell studies for cytotoxicity
- Analyze additional Py<sub>2</sub>N<sub>2</sub> molecules to understand trends in radical scavenging

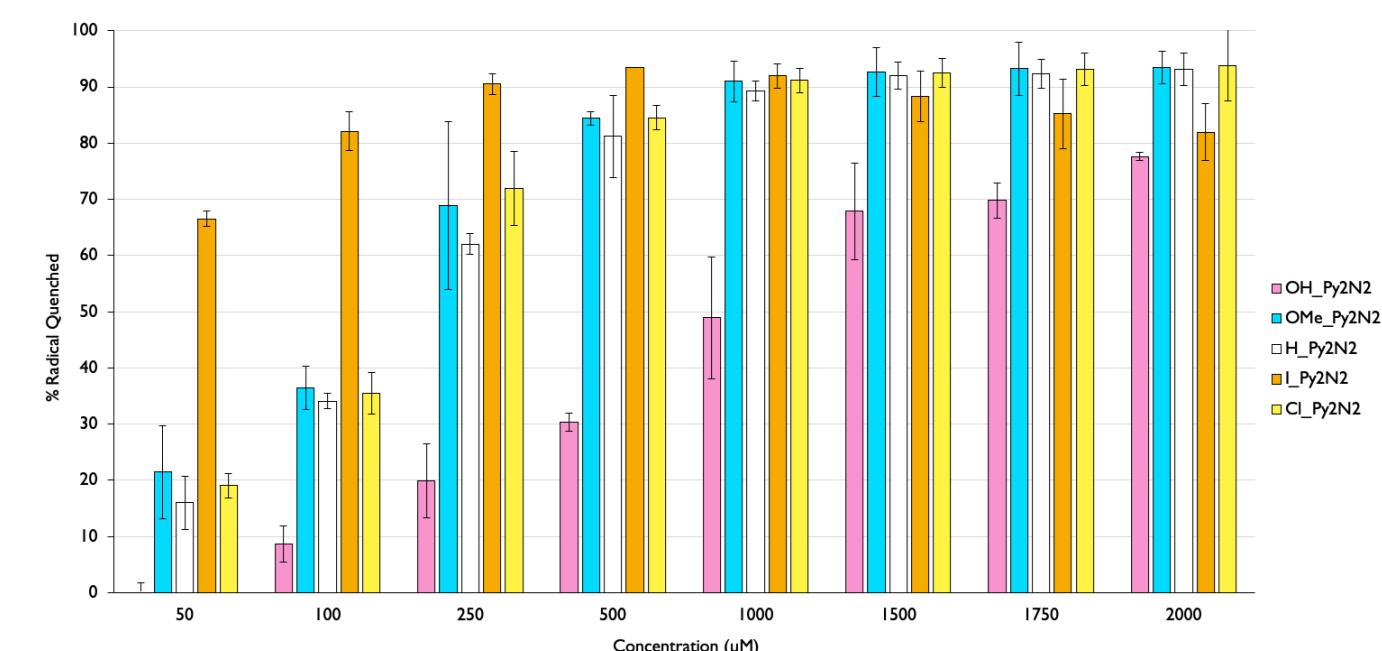
## Py<sub>2</sub>N<sub>2</sub> Series



R = OH, OMe, H, I, Cl



**Design Strategy:** PyN<sub>3</sub> moieties have been shown to provide radical scavenging reactivity. Therefore, adding another pyridine to the ligand backbone should increase scavenging ability.



- ✓ All Py<sub>2</sub>N<sub>2</sub> moieties are strong radical scavengers
- ✓ Quenching activity occurs more effectively at lower concentrations compared to the PyN<sub>3</sub> series
- ✓ Further studies are needed to understand the impact of pyridine ring substitutions on radical scavenging activity within the Py<sub>2</sub>N<sub>2</sub> series
- ✓ **Conclusion:** An additional pyridine group with the macrocyclic core increases radical scavenging activity compared to PyN<sub>3</sub> series

## ACKNOWLEDGEMENTS



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