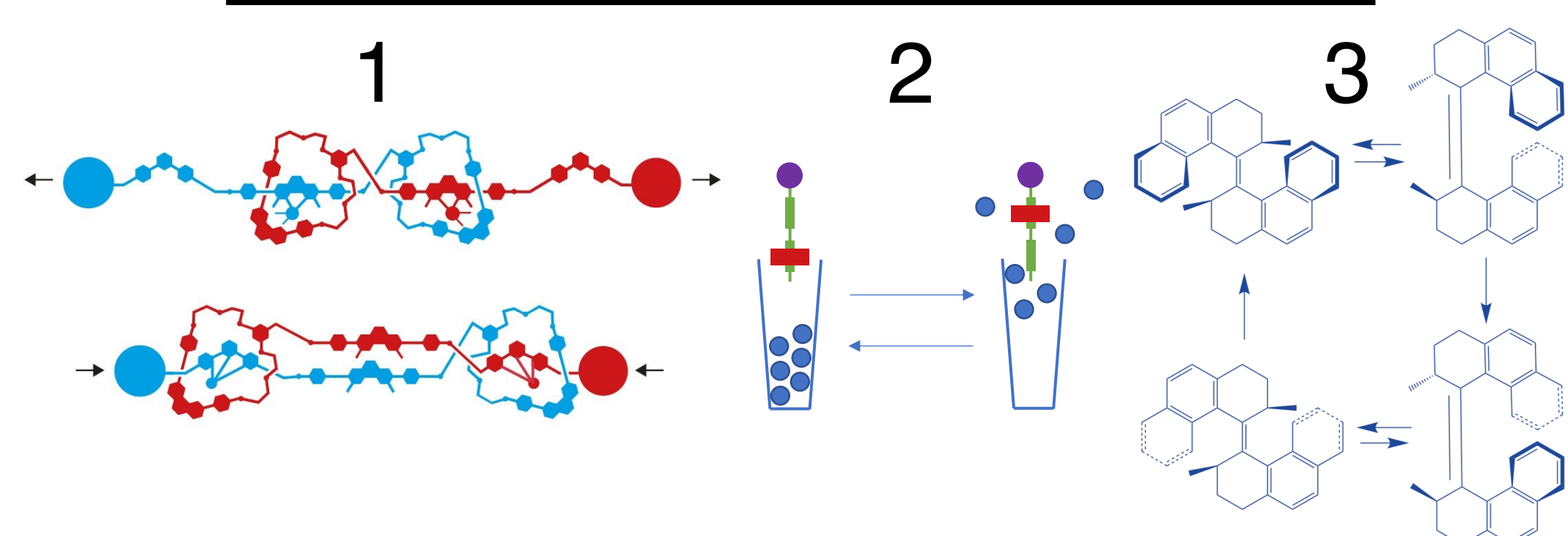


# OPENING THE DOOR ON MOLECULAR HINGES

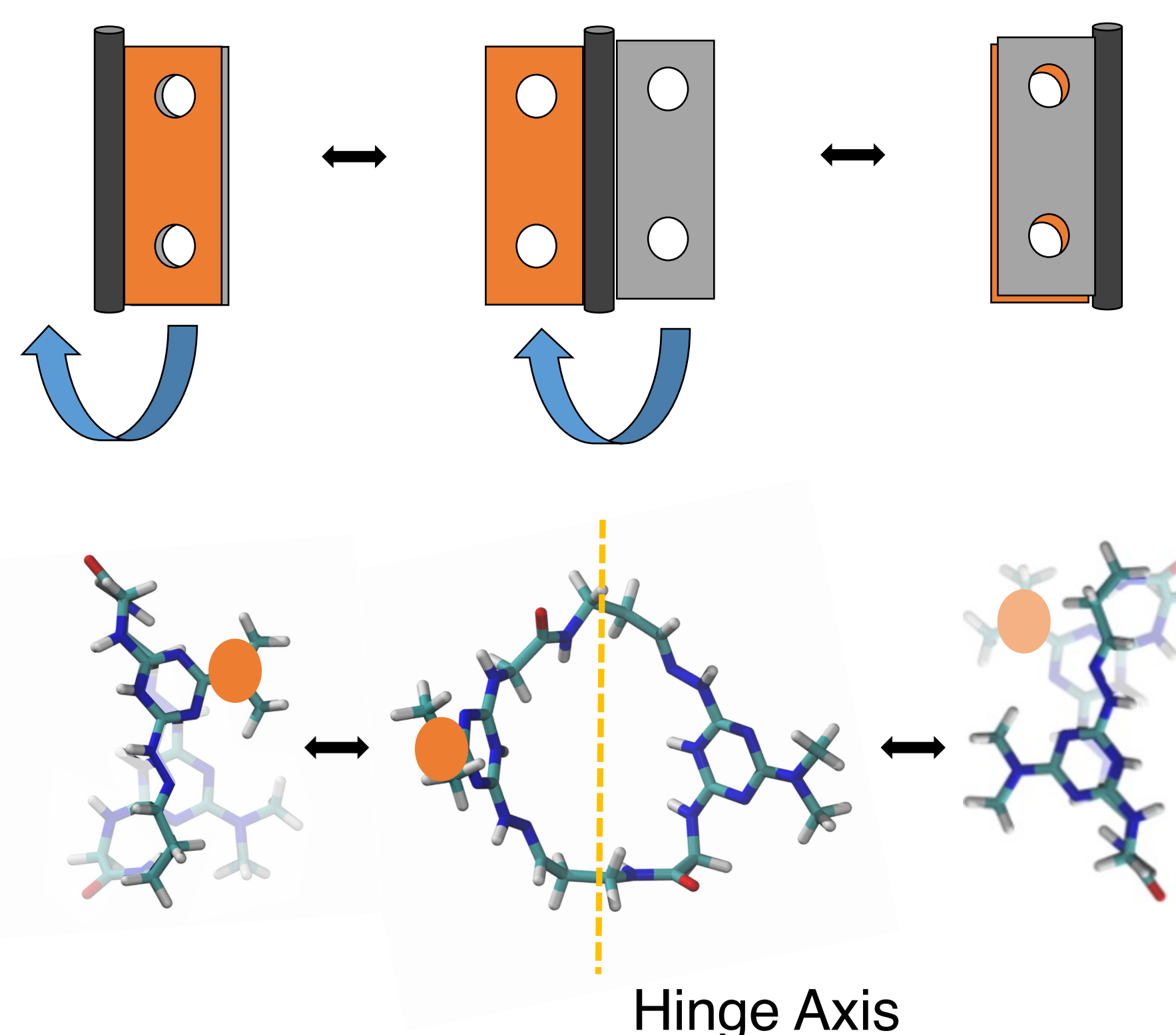
Joseph M. Mellberg, Alexander J. Menke, and Eric E. Simanek\*  
Department of Chemistry & Biochemistry, Texas Christian University

## MOLECULAR MACHINES NOBEL LAUREATES

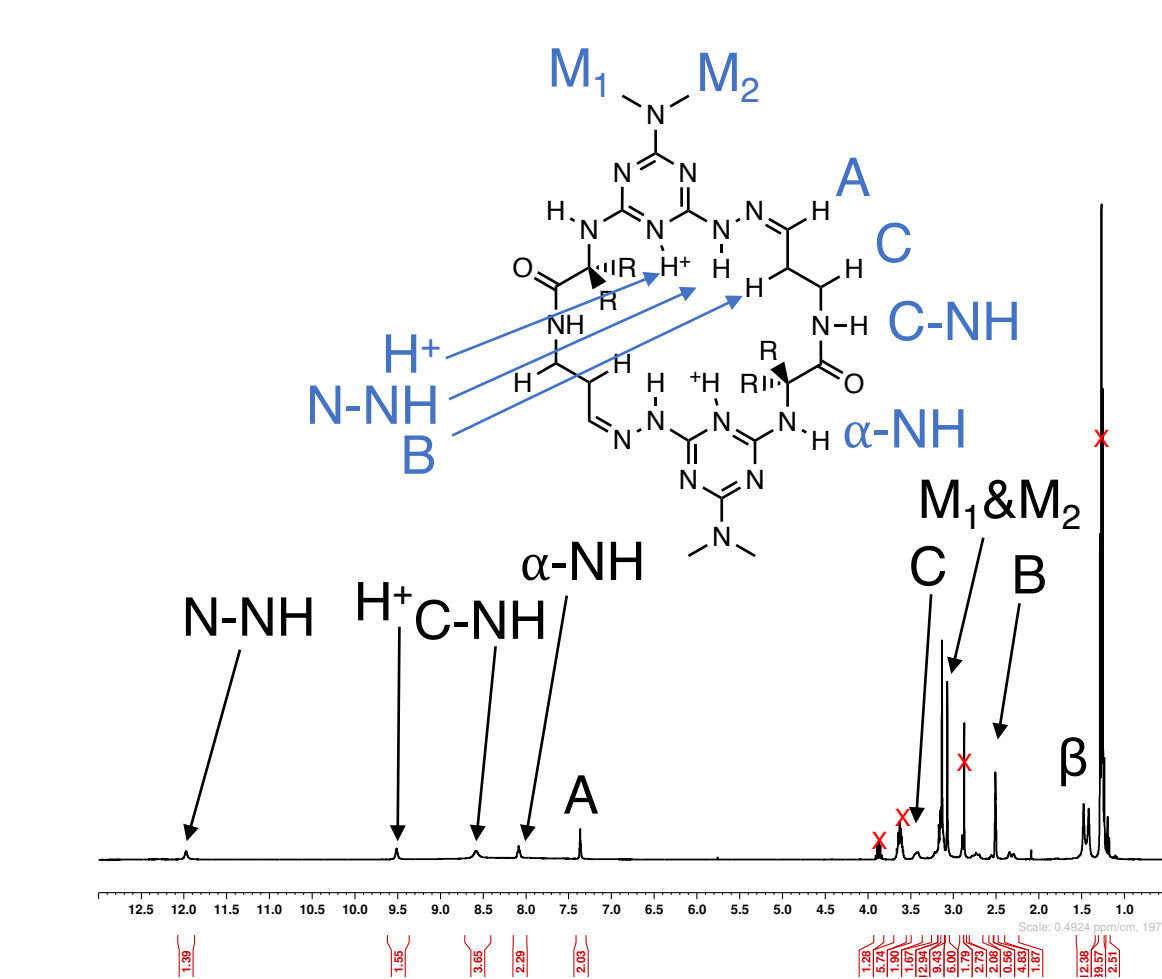


Number	Explanation
1	<ul style="list-style-type: none"> <li><b>Chemist:</b> Jean-Pierre Sauvage (Univ. of Strasbourg)</li> <li><b>Name of Structure:</b> Molecular Muscle</li> <li><b>Description of Activity:</b> Capable of expanding and contracting depending on the type of ion chelated to the molecule.</li> </ul>
2	<ul style="list-style-type: none"> <li><b>Chemist:</b> Sir J. Fraser Stoddart (Northwestern Univ.)</li> <li><b>Name of Structure:</b> Nanovalue</li> <li><b>Description of Activity:</b> Capable of opening or closing depending on the pH of the environment - allowing for precise control of pharmaceutical delivery.</li> </ul>
3	<ul style="list-style-type: none"> <li><b>Chemist:</b> Bernard Feringa (Univ. of Gronigen)</li> <li><b>Name of Structure:</b> Molecular Motor</li> <li><b>Description of Activity:</b> Capable of cycling through different conformations using light and heat.</li> </ul>

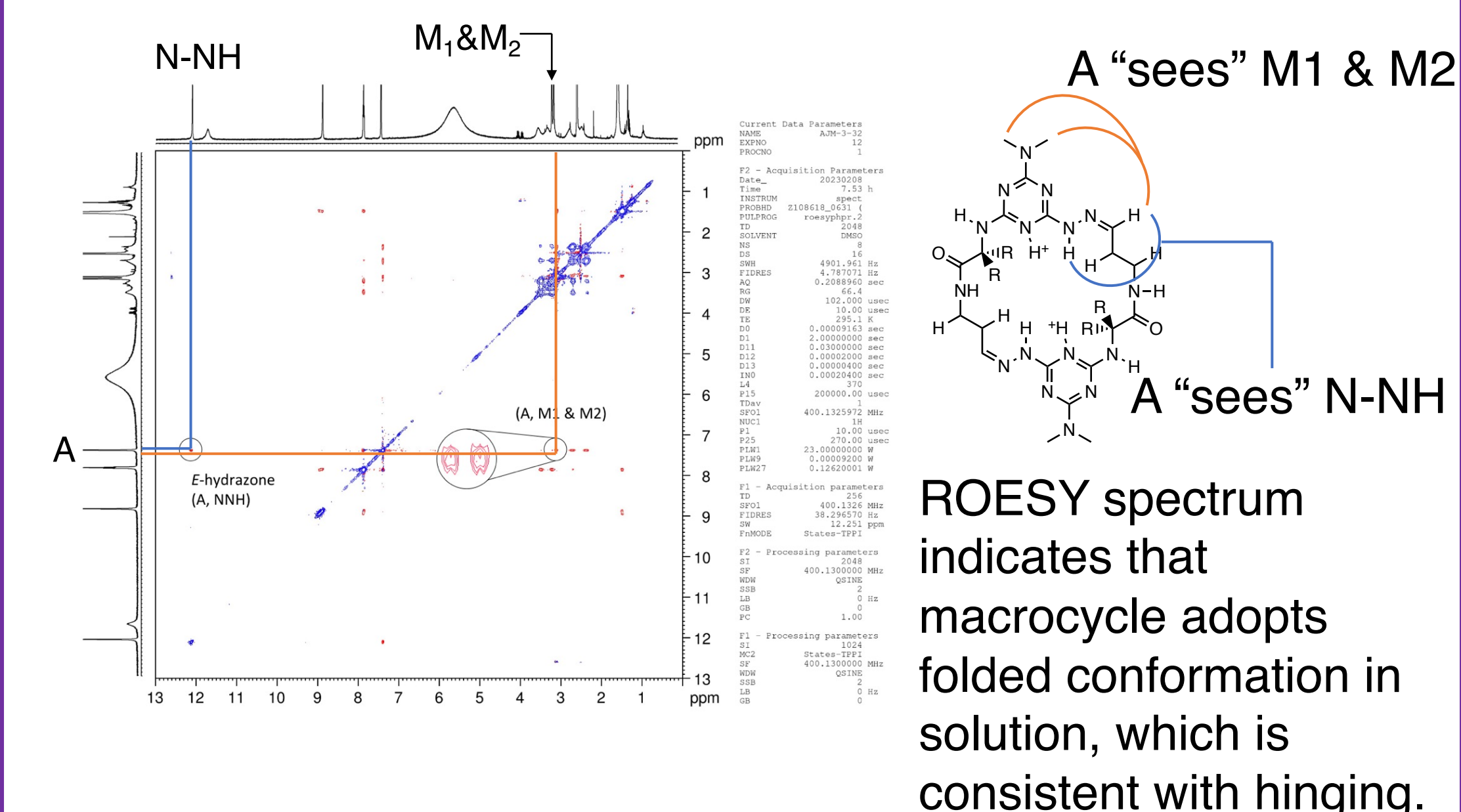
## HINGING MOTION



## MOLECULE IMAGING

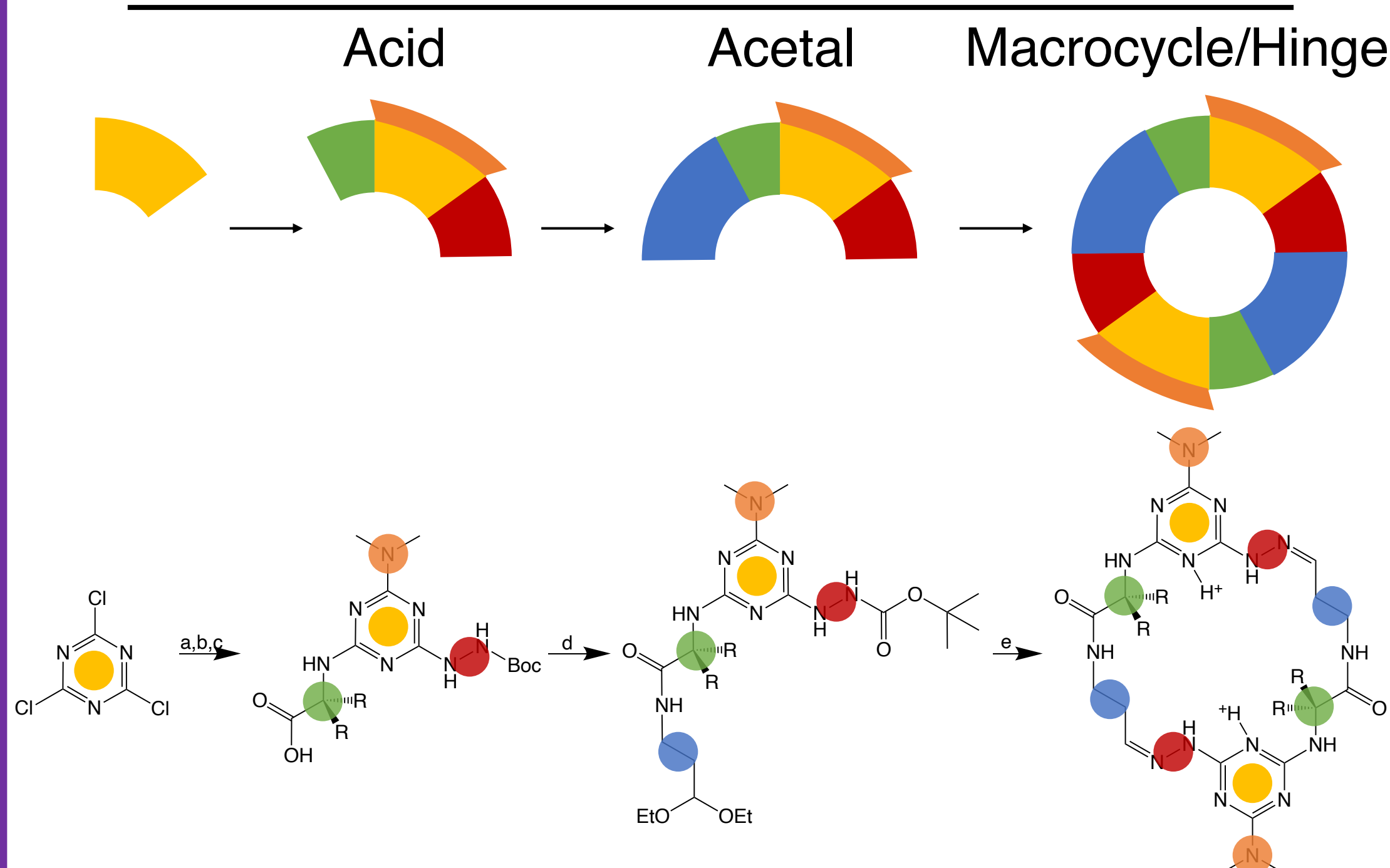


The <sup>1</sup>H NMR spectrum of the **AIB-AIB** macrocycle in DMSO-d<sub>6</sub>. Position of proton A at 7.3 ppm indicates formation of hydrazone, and loss of protecting group indicates complete cyclization.



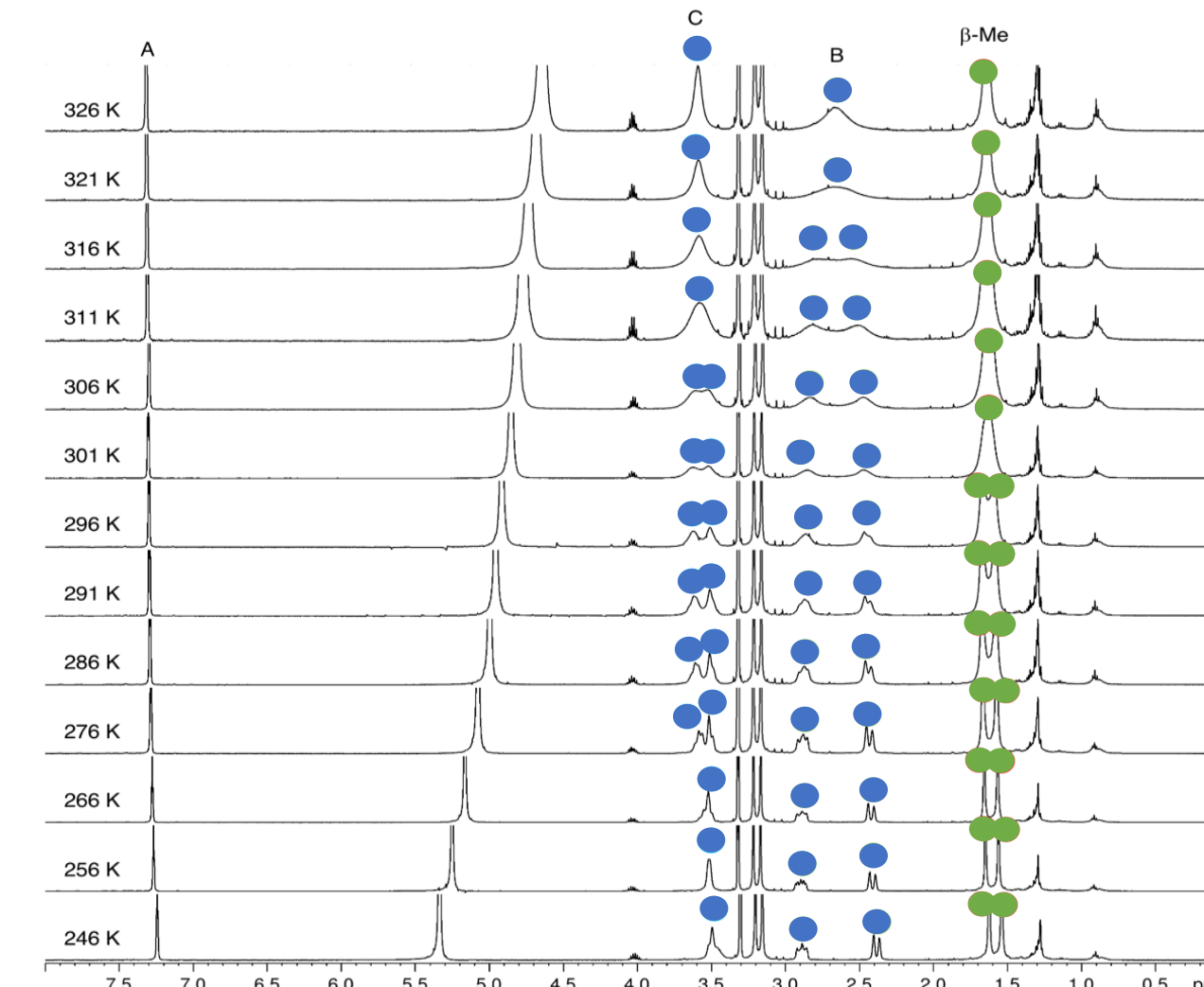
ROESY spectrum indicates that macrocycle adopts folded conformation in solution, which is consistent with hinging.

## SYNTHETIC APPROACH



Letter	Reagents
a	Tert-butyl carbamate (Boc) added over 4.5 hours at -10°C
b	2-aminoisobutyric acid added over 2 hours at 25°C
c	Dimethylamine added over 5 minutes at 25°C
d	HOBT, HBTU, DIPEA added and stirred for 1 hour, then 3,3-diethoxypropylamine added
e	1:1 solution of dichloromethane and trifluoroacetic acid added

## HINGING RATE CALCULATIONS



Variable Temperature NMR in CD<sub>3</sub>OD shows coalescence of the C (blue), B (blue), and β-Me (green) protons.

The rate at which the protons exchange is calculated using **Equation 1**, then the energy required for the hinging motion to occur can be calculated using **Equation 2** (Eyring's equation).

$$k_{exch} \text{ for } \beta\text{-Me: } 106 \text{ sec}^{-1}$$

$$\Delta G^\ddagger \text{ for AIB-AIB: } 15.3 \text{ kcal/mol}$$

$$k_{exch} = \frac{\pi \Delta \nu}{\sqrt{2}}$$

Equation 1

$$\Delta G^\ddagger = -\ln \frac{h k_{exch}}{k_B T_c} * RT_c$$

Equation 2

## CONCLUSION & FUTURE WORK

- 1) Three step process used to successfully and efficiently synthesize **AIB-AIB** macrocycle.
- 2) Characterization of hinging motion possible using various techniques.
- 3) Steric congestion and solvent effects allows for control of hinging motion.
- 4) Measurement of logP and other biological properties will help determine hinging motion's impact on bonding to important intracellular factors.
- 5) Information on bonding and interactions with intracellular elements could inform future therapeutic uses.

## ACKNOWLEDGMENT

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