

Isolating Chiral Nanoparticles from Paper: Templates for Chiral Semiconductor Nanoparticles

Abstract

Perovskites are semiconductor nanomaterial with a cubic crystalline unit that has photoluminescent effects under exciting light^[1]. Perovskites are a new and emerging field of science with various applications, notably most in optoelectronics. New research in this field includes using chiral silica templates for the synthesis of chiral perovskite films.



Figure 1: perovskite unit cell structure with cations such as lead (dark blue), halide atoms at the corners (red) and a central cation such as cesium (orange)



Figure 2: cellulose structure

Introduction

an essential property detected Chirality is throughout nature as well as in synthetic materials with a wide range of applications. At the molecular level, chirality is the ability of two molecules of the same composition to not be structurally superimposable on each other^[3]. However, the necessary conditions required to artificially achieve chiral formation are highly specific. his project began by testing cellulose obtained from several different vendors to determine which product has the ideal properties for use in

chiral films. The use of cellulose as a template porous provides an environmentally alternative friendly templating for chirality onto materials such as silica and eventually perovskites.



chirality 3: Figure in mesoporous silica templated by chiral cellulose





Figure 4: backlit photo nonchiral of cellulose nanocrystals (CNCs) in a separatory funnel at 3 wt%

Figure 5: backlit photo chiral of CNCs in a separatory funnel at 3 wt%



CNCs

- allow slow evaporation.
- Calcinate films to remove cellulose template. Heat to 100°C (2°C/min) and hold for 2 hours. Heat to 540°C (2°C/min) and hold for 6 hours. Slowly cool to room temperature.

2.

3.

4.

- 10.5772/intechopen.98428

Sarafina Gutterres, Jeffery L. Coffer TCU Department of Chemistry and Biochemistry, Texas Christian University, Fort Worth USA

Characterization



Figure 6: nonchiral CNC film 1 after week evaporation at 3 wt%



Figure 7: CNC chiral film 1 week after evaporation at 3 wt%



Figure 8: chiral CNC film 1 week after evaporation at 3 wt%



microscope

Procedure



CNCs and silica film



Use perovskite material to infiltrate silica films.



Silica film



Silica and perovskite film

Conclusion & Future Work

- funnel or in a dried cellulose film
- week)
- chiral formation of CNCs only
- Future work includes:
 - percent
 - infiltration

References

Cava Lab: Perovskite Structure and Derivatives. Princeton University https://www.princeton.edu/~cavalab/tutorials/public/structures/perovskites.html Dey, A. Dey, A.K.; Selection of Optimal Processing Conditino during Removal of Methylene Blue Dye Using Treated Betel Nut Fibre Implementing Desirability Based RSM Approach. Intech Open. 2021. DOI:

Shopsowitz, K., Qi, H., Hamad, W. et al. Free-standing mesoporous silica films with tunable chiral nematic structures. Nature 468, 422–425 (2010). https://doi.org/10.1038/nature09540 Jiang, H. Qu, D. Zou, C. Zheng, H. Xu, Y. Chiral nematic mesoporous silica films enabling multi-colour and on-off switchable circularly polarized luminescence. New J. Chemistry. 2019, 43, 6111. DOI: 10.1039/c9nj00724e

• Two methods for characterizing chiral CNCs are employed: chiral formation in a separatory Both methods elicit multicolor results indicating the presence of chiral CNCs • Time for separation or evaporation must be allowed to achieve chiral formation (at least 1 The conditions for chiral formation in the presence of silica differ from conditions for

 Fine tuning conditions for chiral formation in the presence of silica such as acidity and CNC weight

Testing perovskite materials for